

Material Safety Data Sheet (MSDS) — Natural Latex

Product Name: Natural Latex Concentrate

Supplier / Manufacturer: Basekim Chemical Production Co.

Product Use: Raw latex for manufacturing of gloves, adhesives, foams, coatings,

dipping goods, etc.

Emergency Contact: Basekim Technical Support / Local emergency services

Section 1: Identification

Item	Details
Product Identifier	Natural Latex Concentrate
Synonyms	Natural Rubber Latex, Hevea Latex, Cis-1,4-Polyisoprene Emul <mark>sion</mark>
Material Use	Used in adhesives, foams, rubber dipping, coatings, consumer and medical products
Supplier	Basekim Chemical Production Co., Turkey / UAE
Address / Contact	As per Basekim "Natural Latex" product page

Section 2: Hazard Identification

GHS Classification:

- Skin Sensitization Category 1
- Eye Irritation Category 2A
- Hazardous to Waters Acute Category 2 / Chronic Category 2 (depending on preservative content)
- If ammonia or other stabilizers are present above certain levels, respiratory irritation may occur.

Signal Word: WARNING

Hazard Statements:

- May cause allergic skin reaction.
- Causes serious eye irritation.
- Harmful to aquatic life with long lasting effects.
- May cause respiratory irritation (if inhaled mist or aerosols, or if sufficient ammonia vapour).





Precautionary Statements:

- Avoid breathing vapour / mist.
- Wash hands thoroughly after handling.
- Wear protective gloves, protective clothing, eye protection.
- Avoid release to the environment.
- If in eyes: rinse cautiously with water for several minutes. Remove contact lenses if present and easy to do continue rinsing.

Section 3: Composition / Information on Ingredients

Component	CAS Number	Approxim <mark>ate Co</mark> ncentrati <mark>on</mark>
Natural Rubber (cis-1,4- polyisoprene polymer)	9006-04-6	~ 35-45% (based on "Dry Rubber Content" of ~60%, implying about 60% solid particles; the latex is ~60% solids, 40% water + small non-rubber components)
Water	7732-18-5	~ 35-45%
Ammonia (preservative)	7664-41-7	~0.6-0.8% (for high-ammonia grade)
Proteins, lipids, sugars, resins, minerals (natural minor constituents)	various	~1-2%
Stabilizers / Surfactants / Antioxidants	proprietary / non- hazardous in small %	<1%

Note: Exact formulation may vary depending on grade (High Ammonia, Low Ammonia, Prevulcanized, etc.)

Section 4: First Aid Measures

- Eye Contact: Immediately flush with plenty of water for at least 15 minutes. Remove contact lenses if present and easy to do. Get medical attention if irritation persists.
- **Skin Contact:** Wash skin with soap and water. Remove contaminated clothing. For allergic reaction (rash, redness), get medical advice.
- **Inhalation:** Move to fresh air. If breathing is difficult, provide oxygen. Seek medical attention if symptoms persist.
- **Ingestion:** Rinse mouth. Do NOT induce vomiting unless directed by medical personnel. If swallowed, seek medical attention.





Section 5: Firefighting Measures

- Suitable Extinguishing Media: Water spray, foam, dry chemical, CO₂.
- **Unsuitable Media:** High pressure water jets (may spread latex).
- **Specific Hazards:** Burning latex may emit carbon monoxide, carbon dioxide, possibly ammonia and smoke.
- **Protection of Firefighters:** Use self-contained breathing apparatus (SCBA) and full protective gear.

Section 6: Accidental Release Measures

- **Personal Precautions:** Avoid contact with skin and eyes. Use protective gloves and eye/face protection. Ensure adequate ventilation.
- **Environmental Precautions:** Prevent latex from entering drains or waterways (can cause foaming and environmental harm).
- Methods for Cleanup: Contain spill. Absorb with inert material (e.g. sand, earth). Collect in suitable containers. Wash area with water and detergent; manage wastewater per local regulations.

Section 7: Handling and Storage

- **Handling:** Avoid splashes and generation of aerosols. Use in well-ventilated area. Use personal protective equipment as required.
- Storage: Store between 5°C and 30°C. Avoid freezing. Protect from sunlight and UV. Keep containers sealed to prevent ammonia loss.
- **Incompatibles:** Strong acids, oxidizing agents, heavy metals (which may catalyze degradation).

Section 8: Exposure Controls / Personal Protection

- **Engineering Controls:** Good ventilation. Local exhaust where aerosols/mist is generated.
- **Exposure Limits:**
 - Ammonia vapour (if high enough): see local occupational exposure limits.
 - Natural rubber protein: for sensitized individuals, exposure monitoring may be relevant.
- **Personal Protective Equipment (PPE):**



www.basekim.com



- Respiratory Protection: if spray or mist form, use NIOSH/MSHAapproved respirator.
- o Eye/Face Protection: goggles or face shield.
- o Skin Protection: protective gloves (e.g. nitrile), clothing.
- Hygiene: wash hands before breaks and after work.

Section 9: Physical & Chemical Properties

Property	Typical Value / Range
Appearance	Milky white liquid
Odor	Mild, characteristic; slight ammonia smell (<mark>for high</mark> ammoni <mark>a grade</mark>)
Dry Rubber Content (DRC)	~ 60% ± 2%
рН	9.5 - 11.0
Viscosity	~ 50 - 150 mPa·s (depending on grade <mark>, temp</mark> erature)
Density @ ~25°C	~ 0.94 - 0.97 g/cm³
Coagulum Content	< 0.05%
Volatile Fatty Acids (VFA)	≤ 0.05%
Flash Point	Non-flammable / water-based emulsion (not applicable in standard sense)
Solubility	Dispersible in water (latex is an emulsion), insoluble in organic solvents (natural rubber)

Section 10: Stability and Reactivity

- Chemical Stability: Stable under recommended storage conditions.
- Conditions to Avoid: Freezing, extreme heat (> 40-45°C), UV exposure without stabilizer, strong acid pH < 7, strong oxidizers.
- Incompatible Materials: Strong acids, strong oxidizing agents, heavy metal salts in high concentrations.
- **Hazardous Decomposition Products:** Upon thermal decomposition: CO, CO₂, smoke, possibly ammonia and nitrogen oxides.

Section 11: Toxicological Information

• Acute Toxicity: Natural latex is generally low toxicity.





- **Skin Contact:** Can cause allergic reactions in sensitized individuals; latex protein may trigger contact dermatitis or more serious allergic responses.
- **Eye Contact:** Irritation possible.
- **Inhalation:** Ammonia vapour (high ammonia grades) may cause irritation to respiratory tract. Aerosols may cause irritation.
- Chronic Effects: Repeated skin exposure may lead to sensitization (latex allergy).
- Carcinogenicity / Mutagenicity: Not known to be carcinogenic; but proteins and minor contaminants should be monitored.

Section 12: Ecological Information

- **Ecotoxicity:** Can be harmful to aquatic life, especially due to ammonia and surfactants; foam formation can harm aquatic environment.
- **Persistence and Degradability:** Latex film / solid rubber is not biodegradable; the aqueous emulsion water phase may degrade; polymer particles persist.
- **Bioaccumulative Potential:** Low for soluble components; polymer particles typically not bioavailable.

Section 13: Disposal Considerations

- Dispose in accordance with local, state, and federal regulations.
- Do not discharge into sewers or waterways.
- Clean containers thoroughly before disposal or reuse.

Section 14: Transport Information

- UN Number: Not hazardous under transport regulations (aqueous latex emulsion).
- Proper Shipping Name: Natural Latex Emulsion
- Transport Hazard Class: Non-hazardous for road / sea, unless ammonia vapour concentration or preservative content gives rise to hazard classifications locally.
- Packing Group: Not assigned.

Section 15: Regulatory Information

• Complies with relevant ISO (e.g. ISO 2004 for latex concentrates), ASTM standards (e.g. ASTM D1076), REACH, RoHS as applicable.





Latex proteins may require labelling for allergen risk in some jurisdictions.

Section 16: Other Information

- **Revision Date:** [Insert current date]
- **Prepared by:** Basekim Technical / Quality / Safety Team
- **Disclaimer:** The information provided is based on knowledge to date and intended to describe safety for this product. It is not a guarantee of specific properties. Users must verify suitability for their application and comply with local laws.